NITROGEN BRIDGEHEAD COMPOUNDS III. ** NEW RING TRANSFORMATION I. Zoltán Mészáros and István Hermecz CHINOIN Pharmaceutical and Chemical Works, Budapest

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Lappin¹ was first to observe that when Ia were heated in Dowtherm A at 250 C^O, 2 types of products were formed, depending on the position of the substituent in the pyridine ring: IIa were obtained if the pyridine ring was not substituted in position 6: while 6-substituted pyridines /R=6-Me, 6-OEt/ gave IIIa. Lappin presumed that in the latter case because of the steric hindrance of the 6-R ring-closure on the ring N was prevented and so it took place directly on the carbon atom in position 3. Later, however Antaki² found that cyclization of Ib,c by vacuum distillation even in case R=6-Me, led to the corresponding IIb,c instead of IIIb,c /R=7-Me/. We reported³ that cyclising Ia /R=6-alkyl/ in POCl₃-PPA at 130-40 C^O we obtained IIa /R=6-alkyl/. Under the same conditions Ib /R=6-Me/

$$R = COOC_{2}H_{5}$$

$$COOC_{2}H_{5}$$

$$C_{1}R_{1} = COCC_{2}H_{5}$$

$$C_{1}R_{1} = COC_{2}H_{5}$$

$$C_{1}R_{2} = COC_{2}H_{5}$$

$$C_{1}R_{2} = COC_{2}H_{5}$$

$$C_{1}R_{2} = COC_{2}H_{5}$$

$$C_{2}R_{3} = COC_{2}H_{5}$$

$$C_{1}R_{2} = COC_{2}H_{5}$$

$$C_{2}R_{3} = COC_{2}H_{5}$$

$$C_{3}R_{4} = COC_{4$$

decomposed and Ic /R=6-Me/ failed to react. When the course of the cyclization of Ia /R=6-Me/ performed in Dowtherm A was followed by TLC /Kieselgel G, PhH-MeOH=4:1/, between the spots of Ia /R=6-Me, Rf=0,81/ and the formed IIIa /R=7-Me, Rf=0,39/ a third spot Rf=0,53 identified unambiguously as IIa /R=6-Me/ was detected on the cromatogram. On slow heating of the reaction mixture the

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spot of IIa /R=6-Me/ appeared first /at 160-180 $C^{\circ}/$, and that of IIIa /R=7-Me/ became visible at a higher temperature only /180-200 $C^{\circ}/$.

Performing cyclization of Ib,c /R=6-Me/ in paraffin oil at 300 C^O, in contrast to Antaki, we failed to obtain the corresponding IIb,c /R=6-Me/, but obtained IIIb $\left[R=7-\text{Me}; \text{mp}:\right>300 \text{ C}^{\text{O}} / \text{DMF}/; Y:80\$; \text{UV /EtOH/ 32lnm /lg \mathcal{E} 4,07/; 261 /4,24/; 255i /4,20/] and IIIc <math>\left[R=7-\text{Me}; \text{mp}:\right>300 \text{ C}^{\text{O}} / \text{DMF}/; Y:60\$; \text{UV /EtOH/: 327 /4,02/}; 318i /3,98/; 254,5 /4,24/; 247 /4,19/] /In the reaction of Ib /R=6-Me/ performed in Dowtherm A after 2 hours IIIb /R=7-Me/ was obtained in 65 \$ yield, while Ic /R=6-Me/ failed to react./$

The above facts lead to the conclusion that cyclization of I in the first step always occurs on the ring N, i.e. II are formed which under the given conditions may andergo thermic isomerization and tranform to III by 1-3 N-C acyl migration Acyl migration is rendered possible by the fact that the 6-R and the 4-oxo-group of II /6-R \neq H/ are practically in the same plane and as a consequence of it C_4 -N₅ bond subjected to strong strain, splits up readily upon the attact of suitable energy. /According to the X-ray investigations the C_4 -N₅ bond length in IIa /R=6-Me/ is 1,472Å./

Corresponding to the above theoretical consideration, IIa /R=6-Me/ prepared in $POCl_3$ -PPA reaction³ could be transformed to IIIa /R=7-Me, Y:80%/ by heating in Dowtherm A. II /R=H/ could not be isomerized thermically.

The presence of sterically large substituents /e.g. Br/ in position 6 of Ia appears to prevent ring closure. Thus, our earlier endeavours to cyclisize Ia /R=6-Br/ in POCl₃/PPA failed. Ia /R=6-Br/ was regained unchanged from the reaction mixture. Upon heating in Dowtherm A the compound decomposed 1.

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